Nuclear Magnetic Resonance Spectrum of Phytoene

Ginzaburo Suzue, Katsuo Tsukada and Shozo Tanaka

Department of Chemistry, Faculty of Science, Kyoto University, Sakyo-ku, Kyoto

(Received February 20, 1967)

Phytoene, a colorless polyene, occurs widely in nature¹⁾ and is thought to play an important role in the biosynthesis of carotenoids.^{2,3)} Its structure is considered to be 2, 6, 10, 14, 19, 23, 27, 31-octamethyldotriacontanonaene - 2, 6, 10, 14, 16, 18, 22, 26, 30, mainly on the basis of its chemical properties.4) The NMR spectrum has been reported by Davis et al,5) and by Jungalwala et al.6) However, only the values for the methyl groups were reported in the former paper,50 whereas the chart

1) W. J. Rabourn and F. W. Quackenbush, Arch.

Biochem. Biophys., **44**, 159 (1953).
2) J. W. Porter and D. G. Anderson, *ibid.*, **97**, 520 (1962).

S. L. Jensen, Annual Rev. Microbiol., 19, 163

(1965).
4) W. J. Rabourn and F. W. Quackenbush, Arch. Biochem. Biophys., 61, 111 (1956).

5) J. B. Davis, L. M. Jackman, P. T. Siddons and B. C. L. Weedon, *Proc. Chem. Soc.*, 261 (1961).
6) F. B. Jungalwala and J. W. Porter, *Arch. Biochem. Biophys.*, **110**, 291 (1965).

shown in the latter paper has peaks in the region of τ 8.5—9.3 which are unexplainable on the basis of the structure reported4); besides, the relative intensities of the respective peaks are not given. The NMR spectrum has, accordingly, been remeasured on a pure sample of phytoene in order to check the validity of the published spectrum and the structure.

The method of the extraction and purification of phytoene from tomatoes was practically the same as that previously reported,69 but two additional preparative thin-layer chromatographic separations (plate: 20 cm × 20 cm, silica gel G activated for 2 hr at 110°C, thickness: 1.0 mm, solvent: hexane-ether, 99-1) were carried out. The position of phytoene was determined by spraying the plate with a 0.5% solution of potassium permanganate after covering it with a slitted cardboard (slits: 0.3 mm with 3-cm intervals). The position of the plate was marked, and the

silica gel on that portion was scraped off and eluted with freshly-distilled ether-hexane (50:50). Phytoene, obtained by the removal of the solvent in vacuo, was used for the measurement.

The NMR spectrum of phytoene was run in deuterochloroform (concentration 10 mg/0.5 ml), with tetramethylsilane as internal reference, using a

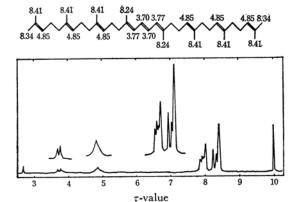


Fig. 1. NMR spectrum and its assignment of phytoene in CDCl₃ at 60 Mc/sec.

JEOL 60 Mc 3H-60 spectrometer. In the following the observed τ values (see Fig. 1) are recorded, together with the relative numbers of protons observed (obtained by electronic integration) and the relative numbers of protons expected (shown in parentheses): 3.70, central -CH= 1.9 (2); 3.77, -CH= adjacent to central unsaturation 1.9 (2); 4.85, other -CH=5.7 (6); 7.85-8.02 $-CH_2-$ 24.2 (24); 8.24, central -CH₃ 6.0 (6); 8.34, terminal -CH₃ (cis to H) 6.0 (6); 8.41 other -CH₃ (trans to H) 18.3 (18). These assignments are based on the spectra of the polyisoprenoid chains in naphtoquinone7) and carotenoids,5,8) and also on a cyclic triene.9) In the present experiment, the fine structure of the -CH= protons was not well resolved, possibly because of the large size of the phytoene molecule. However, the NMR spectrum is in good agreement with the structure of phytoene.

⁷⁾ R. H. Baum and M. I. Dolin, J. Biol. Chem., 240, 3425 (1965).

⁸⁾ L. Ryvarden and S. L. Jensen, *Acta Chem. Scand.*, **18**, 643 (1964).

⁹⁾ NMR Spectra Catalog, No. 158, Varian Associates.